

1st Network Meeting: ESF RNP CASIMIR

**November 29-30, 2008 at the
Abbey of Royaumont, France**

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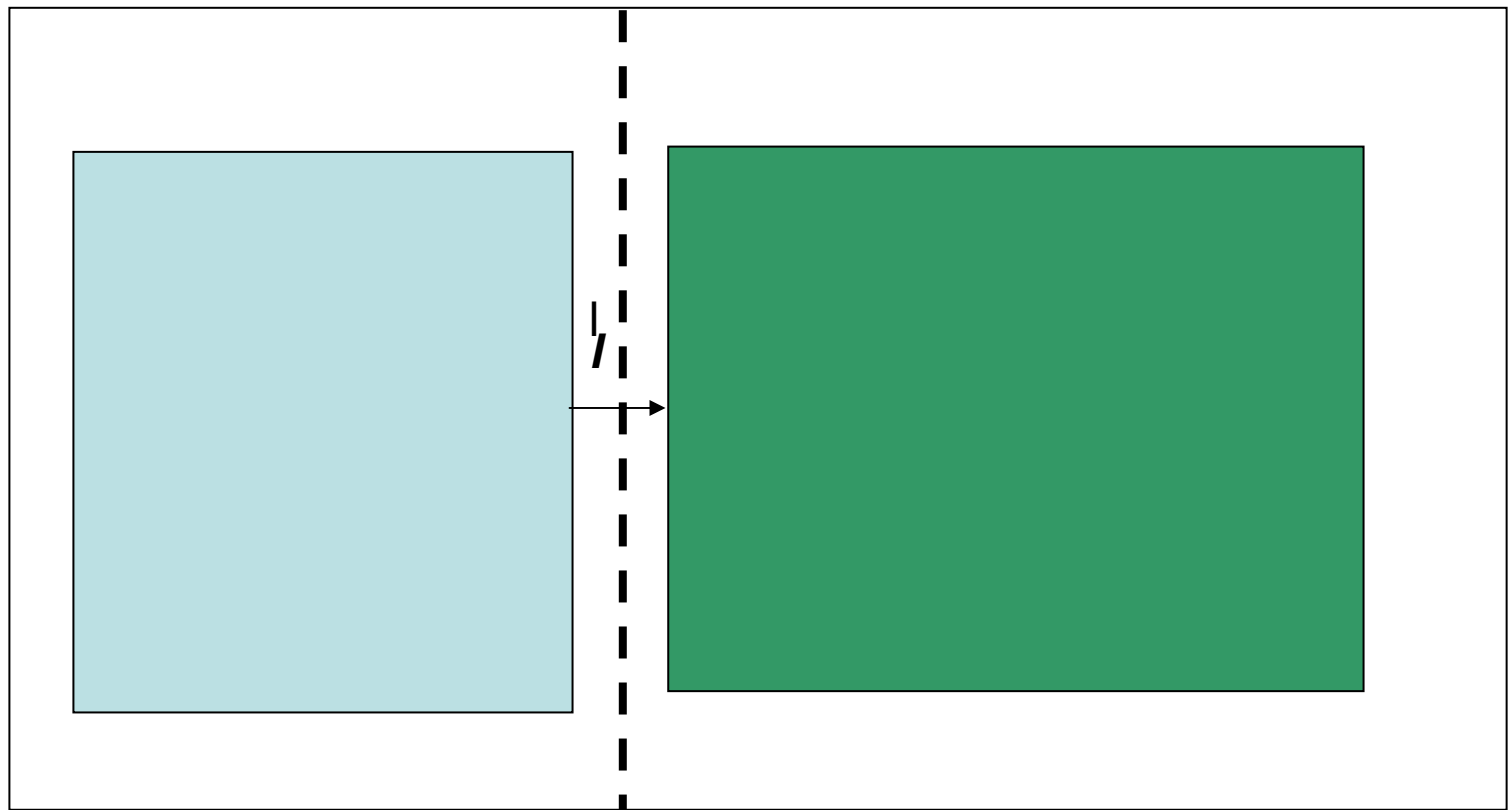
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Part 1. Casimir-Lifshitz Forces and Entropy

Interaction between two bodies



$$f = \sigma_{xx}$$

What theory can say about Entropy?

The theory gives the ONLY one quantity

- the force : $f(T, l)$. Thermodynamics :

$$f(T, l) = - \left(\frac{\partial F(T, l)}{\partial l} \right)_T$$

$$\left(\frac{\partial S(T, l)}{\partial l} \right)_T = \left(\frac{\partial f(T, l)}{\partial T} \right)_l$$

$$\Delta S = S(T, l) - S(T, l_0) = \int_{l_0}^l \left(\frac{\partial f(T, l)}{\partial T} \right)_l dl$$

Nernst's theorem

According to Nernst's theorem
(it is valid for ordered bodies -
quantum fluids, ideal crystals):

$$S(T, l) |_{T \rightarrow 0} \rightarrow 0, T \rightarrow 0$$

Consequences of Nernst's theorem

It MUST be

$$\left(\frac{\partial S(T, l)}{\partial l} \right)_T = \left(\frac{\partial f(T, l)}{\partial T} \right)_l \rightarrow 0$$

$$T \rightarrow 0, l = \text{const}$$

$$\Delta S \equiv S(T, l) - S(T, l_0) \propto \left(\frac{\partial f(T, l)}{\partial T} \right)_l \rightarrow 0$$

No information about signs at finite T !

Is it important to keep $l = \text{const}$? **Yes!**

Instructive example. Degenerate ideal gas.

Entropy per atom :

$$S(T, V) = S\left(TV^{2/3} / N^{2/3}\right)$$

Let : $T \rightarrow 0$ **but**

$$V \rightarrow \infty, TV^{2/3} \rightarrow \text{const}$$

Then $S(T, V) \neq 0$ at $T \rightarrow 0$

$l \rightarrow \infty$ means $V \rightarrow \infty$!!!

Popular definition of “Casimir entropy”

$$l_0 \rightarrow \infty$$

$$S_C(T, l) \equiv - \int_l^\infty \left(\frac{\partial f(T, l)}{\partial T} \right)_l dl$$

$$= S(T, l) - S(T, \infty)$$

$$S_C(T, l) > 0 ?$$

$$T \rightarrow 0 : S_C(T, l) |_{l \rightarrow \infty} \rightarrow -S(T, \infty) \rightarrow 0 ?$$

Why? No obligations.

About the sign of “Casimir entropy”

$$S_C(T, l) \equiv S(T, l) - S(T, \infty)$$

A point : temperature dependences of

$$S(T, l) \text{ and } S(T, \infty) \text{ at } T \rightarrow 0$$

can be different.

Large l can give finite contribution.

If at $T \rightarrow 0 : |S_C(T, l)| \gg |\Delta S(T, l, l_0)|,$

$$S_C(T, l) \approx -S(T, \infty) < 0$$

S_C is negative just due to Nernst's theorem!

When “Casimir entropy” can be negative ?

Let at large l and small T

the main contribution in f

is from $\omega \sim \omega_0(T, l)$. If at $l \rightarrow \infty$

$\hbar\omega_0 \leq k_B T$, $S(T, \infty)$ will be large

and even finite at $T = 0$.

Then $S_C < 0$.

Conclusion:

The “Casimir entropy” S_C is not the simplest characteristic of thermal properties of the Casimir-Lifshitz forces.

Examples

$$T \rightarrow 0$$

1. Ideal metal. Anomalous skin effect theory.

$$\frac{\partial S}{\partial l} \sim \Delta S \propto T, S_C \propto T^{2/3}$$

$$|S_C| \gg |\Delta S|, S_C < 0$$

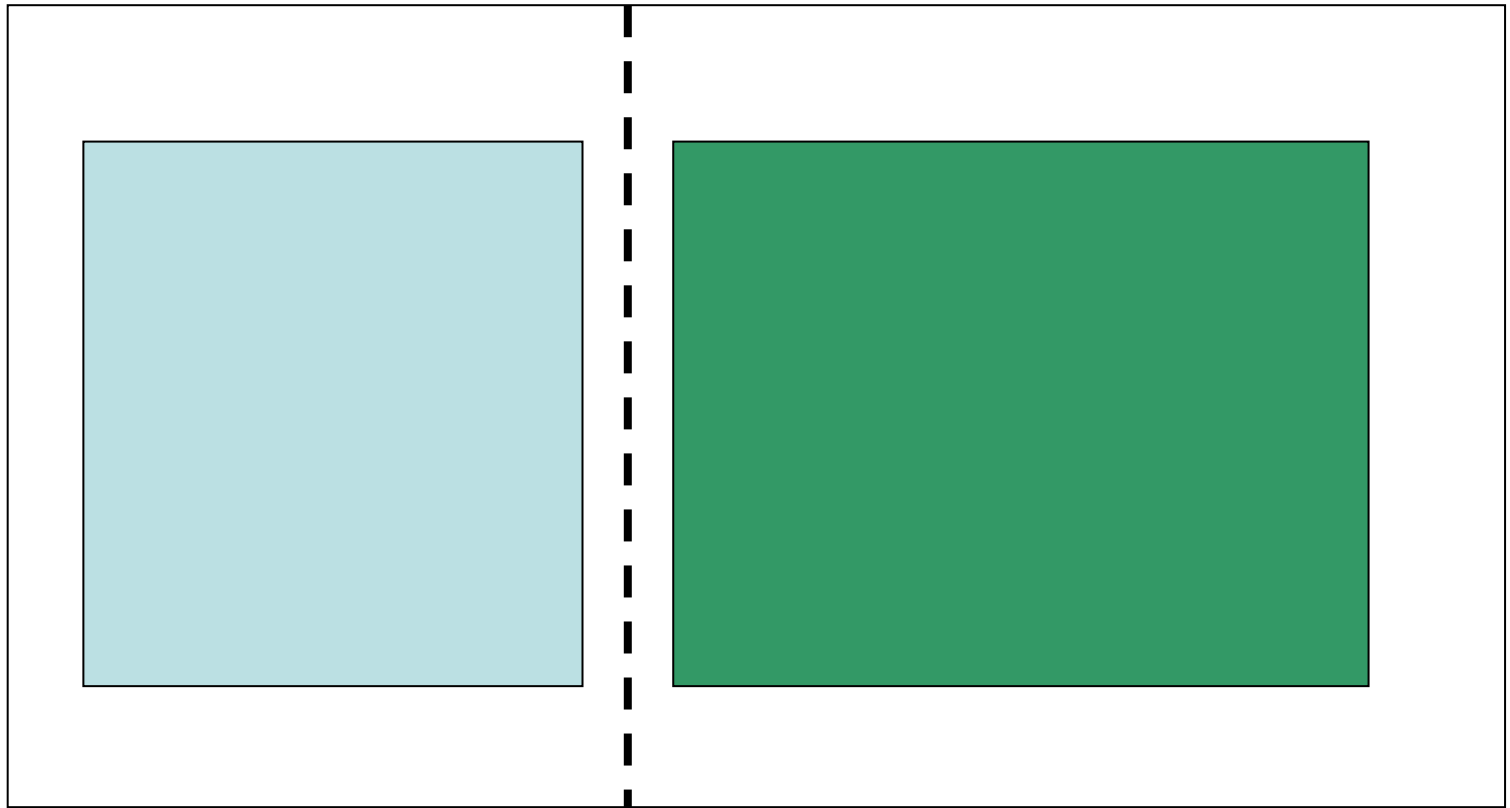
2. Carriers localized by disorder.

$$\frac{\partial S}{\partial l} \sim \Delta S \sim S_C \rightarrow \text{const}$$

Sign S_C cannot be predicted. (Actually $S_C > 0$.)

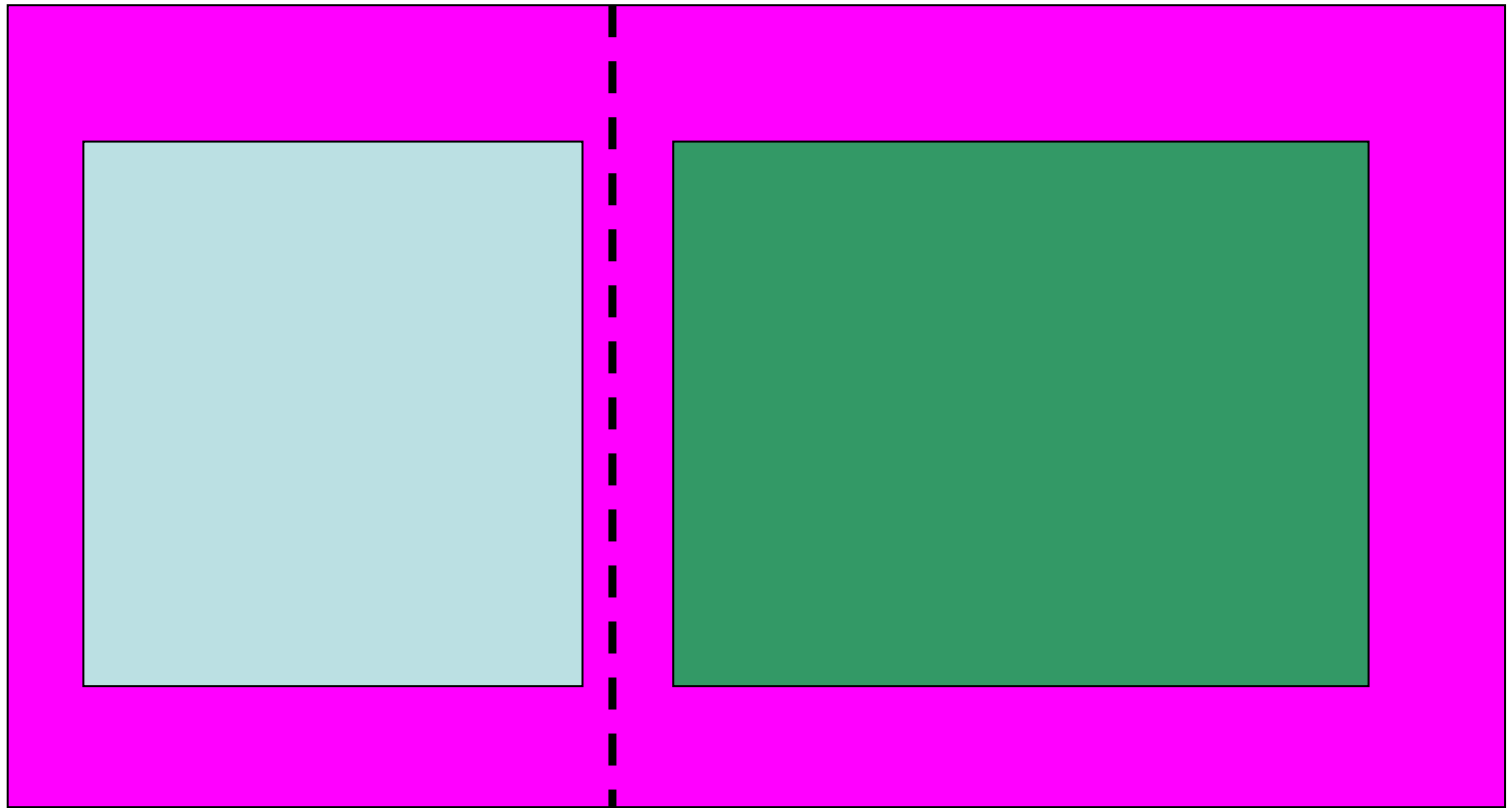
Part 2. Interaction through medium

Interaction through vacuum



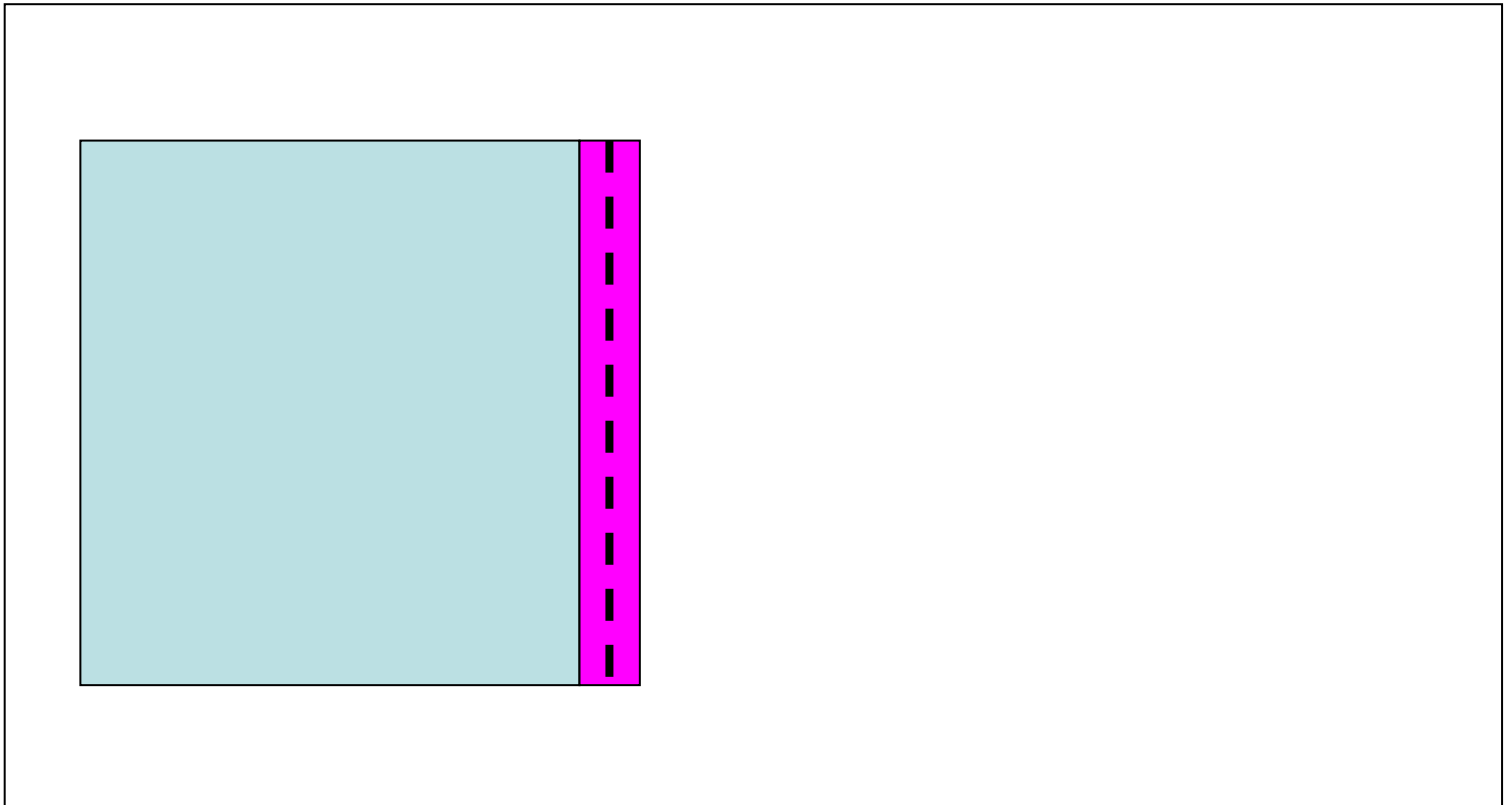
σ_{xx}

Interaction through medium



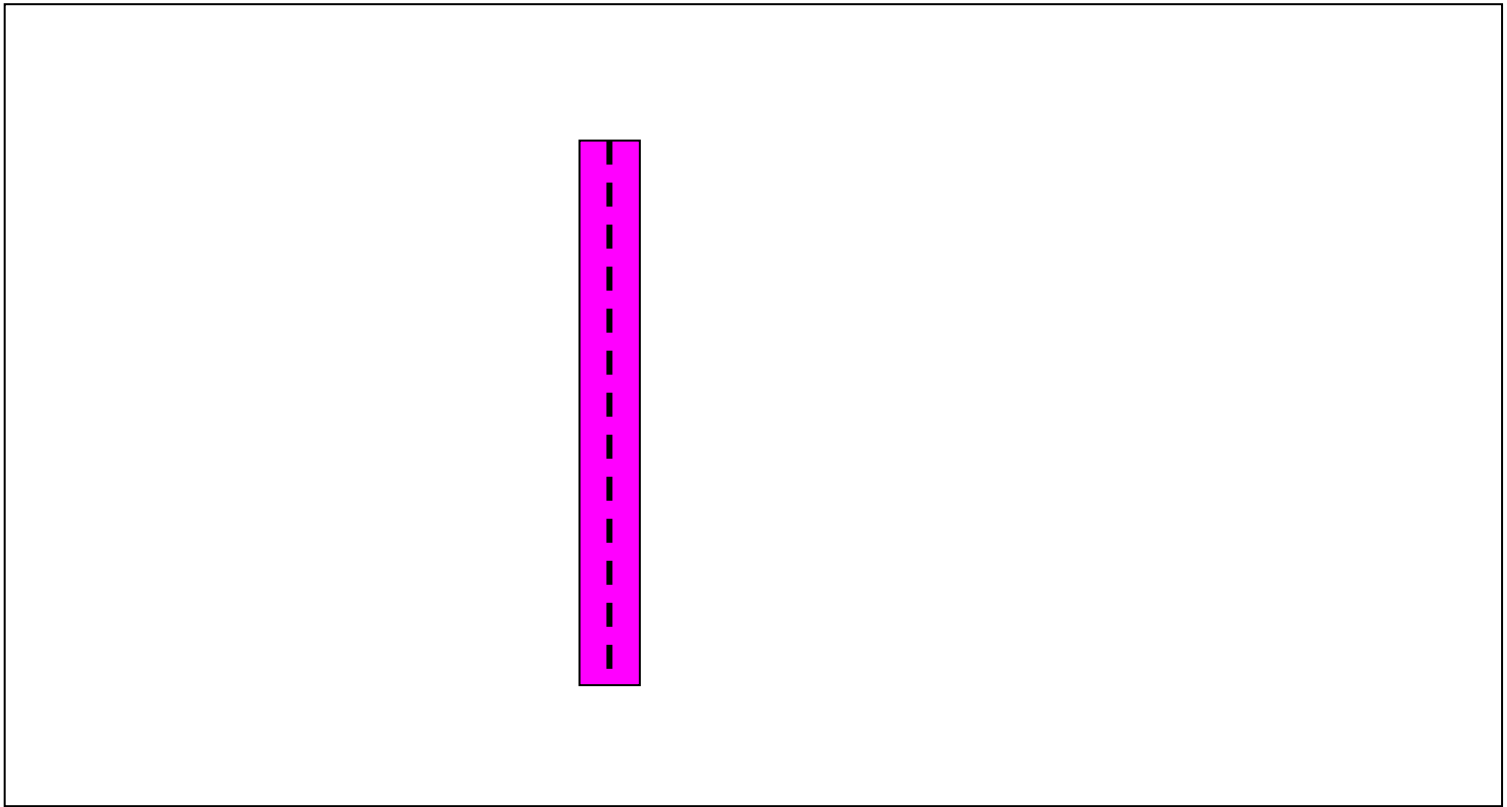
$$\sigma_{xx} = ?$$

Problem of a film



$$\mu(d) = \sigma_{xx}(d)$$

Problem of a film in vacuum



$$\mu(d) = \mu_{\infty} + \delta\mu(d)$$

Generalization to electrolytes?